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Key indicators

Single-crystal X-ray study
 T = 293 K
 Mean $\sigma(\text{C}-\text{C}) = 0.002 \text{ \AA}$
 R factor = 0.030
 wR factor = 0.084
 Data-to-parameter ratio = 13.3

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

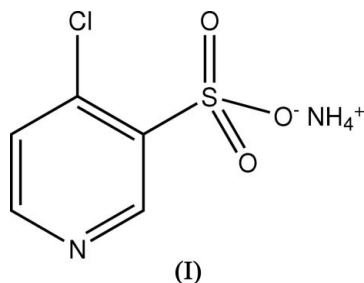
Ammonium 4-chloropyridine-3-sulfonate

The title compound, $\text{NH}_4^+ \cdot \text{C}_5\text{H}_3\text{ClNO}_3\text{S}^-$, was prepared by the hydrolysis of 4-chloropyridine-3-sulfonamide. In the crystal structure, a three-dimensional network is formed *via* $\text{N}-\text{H} \cdots \text{O}$ [$\text{H} \cdots \text{O} = 1.97(3) - 2.41(2) \text{ \AA}$] and $\text{N}-\text{H} \cdots \text{N}$ [$\text{H} \cdots \text{N} = 2.13(3) \text{ \AA}$] hydrogen bonds.

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Comment

The title compound, (I), is a key intermediate of in the synthesis of torsemide (Che *et al.*, 2005), a loop diuretic, which has been found to be effective in the treatment of edema associated with chronic renal failure.



The molecular structure of (I) is shown in Fig. 1, selected torsion angles are given in Table 1 and details of the hydrogen-bonding geometry are given in Table 2 and Fig. 2.

Experimental

4-Chloropyridine-3-sulfonamide (3.0 g) was dissolved in water (20 ml) and stirred for 15 min. Colorless crystals of the title compound suitable for X-ray diffraction analysis were obtained after two weeks (m.p. 499–501 K).

Crystal data

$\text{NH}_4^+ \cdot \text{C}_5\text{H}_3\text{ClNO}_3\text{S}^-$	$D_x = 1.655 \text{ Mg m}^{-3}$
$M_r = 210.64$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 2851 reflections
$a = 8.3501(6) \text{ \AA}$	$\theta = 4.9 - 56.4^\circ$
$b = 7.6684(6) \text{ \AA}$	$\mu = 0.67 \text{ mm}^{-1}$
$c = 13.3162(10) \text{ \AA}$	$T = 293(2) \text{ K}$
$\beta = 97.6170(10)^\circ$	Prism, colorless
$V = 845.14(11) \text{ \AA}^3$	$0.50 \times 0.48 \times 0.24 \text{ mm}$
$Z = 4$	

Data collection

Bruker SMART CCD diffractometer	1835 independent reflections
φ and ω scans	1628 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Bruker, 2000)	$R_{\text{int}} = 0.033$
$T_{\text{min}} = 0.719, T_{\text{max}} = 0.850$	$\theta_{\text{max}} = 27.0^\circ$
4804 measured reflections	$h = -10 \rightarrow 10$
	$k = -9 \rightarrow 9$
	$l = -16 \rightarrow 15$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.030$
 $wR(F^2) = 0.084$
 $S = 1.07$
 1835 reflections
 138 parameters
 All H-atom parameters refined

$w = 1/[\sigma^2(F_o^2) + (0.0476P)^2 + 0.1484P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.40 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.32 \text{ e } \text{\AA}^{-3}$
 Extinction correction: *SHELXL97*
 Extinction coefficient: 0.042 (3)

Table 1

Selected torsion angles ($^\circ$).

C5–N1–C1–C2	0.6 (3)	S–C2–C3–C1	1.5 (2)
N1–C1–C2–C3	–0.9 (3)	C2–C3–C4–C5	–0.3 (3)
N1–C1–C2–S	178.43 (13)	C1–C3–C4–C5	179.66 (14)
C1–C2–C3–C4	0.7 (2)	C1–N1–C5–C4	–0.1 (3)
S–C2–C3–C4	–178.61 (13)	C3–C4–C5–N1	0.0 (3)
C1–C2–C3–Cl	–179.25 (12)		

Table 2

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N2–H2 \cdots O2	0.78 (3)	2.16 (3)	2.894 (2)	159 (2)
N2–H3 \cdots O2 ⁱ	0.85 (2)	2.18 (2)	2.929 (2)	148 (2)
N2–H3 \cdots O3 ⁱⁱ	0.85 (2)	2.41 (2)	2.916 (2)	118.9 (19)
N2–H6 \cdots N1 ⁱⁱⁱ	0.86 (3)	2.13 (3)	2.944 (2)	156 (2)
N2–H7 \cdots O1 ^{iv}	0.91 (3)	1.97 (3)	2.877 (2)	173 (2)

Symmetry codes: (i) $-x + 2, y + \frac{1}{2}, -z + \frac{3}{2}$; (ii) $x, y + 1, z$; (iii) $-x + 1, -y + 1, -z + 1$; (iv) $-x + 2, -y + 1, -z + 1$.

All H atoms were refined independently with isotropic displacement parameters [$C-H = 0.90(2)–0.937(19) \text{\AA}$ and $N-H = 0.78(3)–0.91(3) \text{\AA}$].

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2000); software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2003).

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Bruker (2000). *SMART, SAINT, SADABS* and *SHELXTL*. Bruker AXS inc., Madison, Wisconsin, USA.

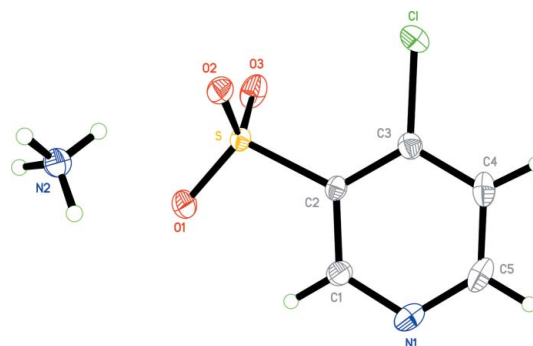


Figure 1

The structure of (I), showing displacement ellipsoids drawn at the 40% probability level. H atoms are represented by circles of arbitrary size.

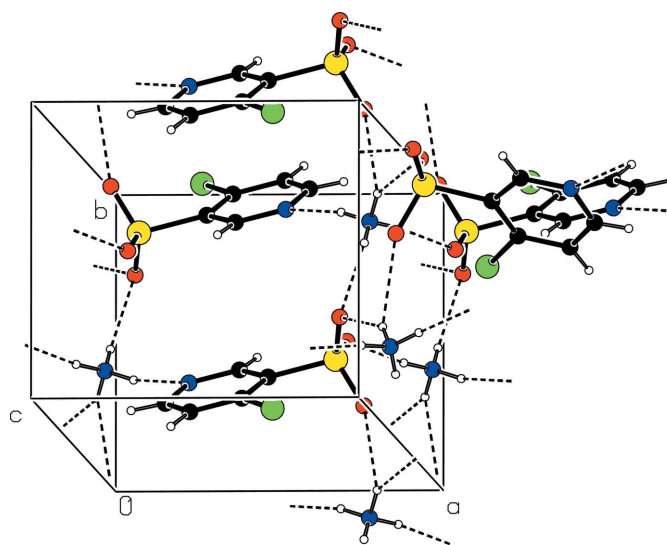


Figure 2

Partial packing plot (Spek, 2003), showing hydrogen bonds as dashed lines. Color codes: green Cl, yellow S, red O, blue N and black C.

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